

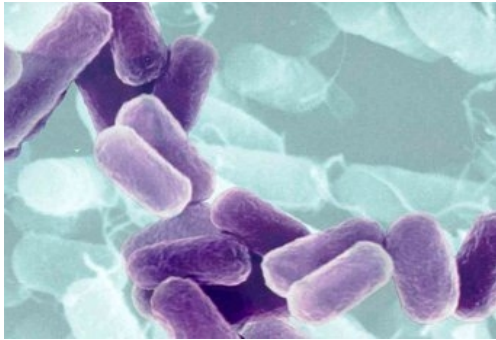


Metabolic networks

Wolfram Liebermeister

Humboldt-Universität zu Berlin – Theoretische Biophysik

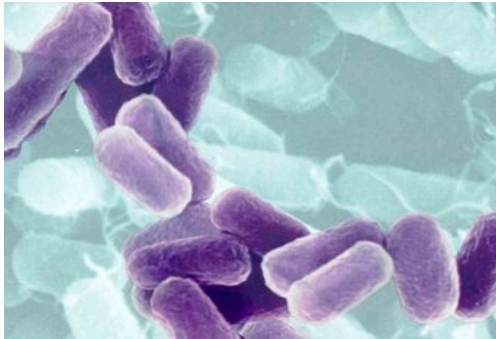
How can a living being emerge just from sugar, water, and a couple of salts?



Minimal Medium for *E. coli*

Glucose	5 g/l
Na ₂ HPO ₄	6 g/l
KH ₂ PO ₄	3 g/l
NH ₄ Cl	1 g/l
NaCl	0.5 g/l
MgSO ₄	0.12 g/l
CaCl ₂	0.01 g/l

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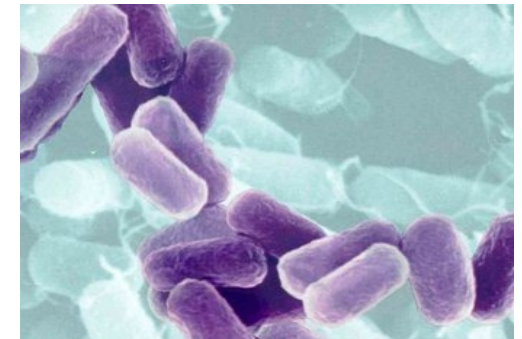
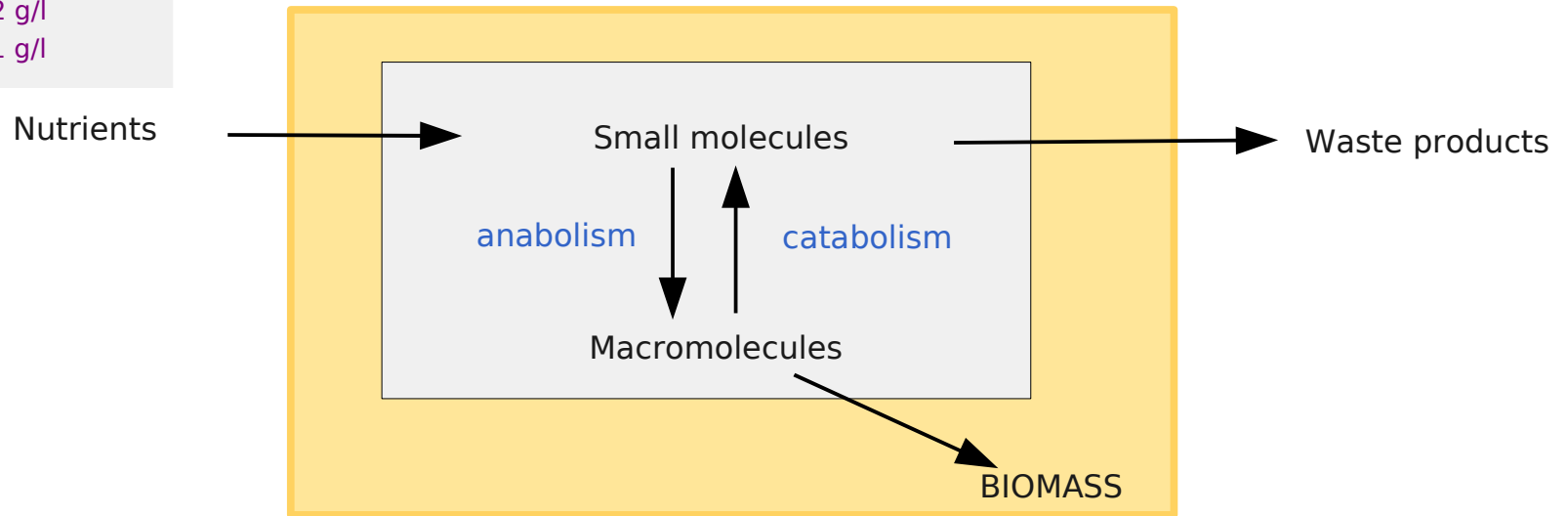
L'essentiel est invisible pour les yeux.



Metabolic networks produce materials and energy for the cell

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Overview

What are metabolic networks and how do they work ?

How can we use models to understand their dynamics ?

How can we predict fluxes in large networks ?

How do metabolic systems respond to perturbations ?

What standards, resources, and software are available ?

Metabolic networks

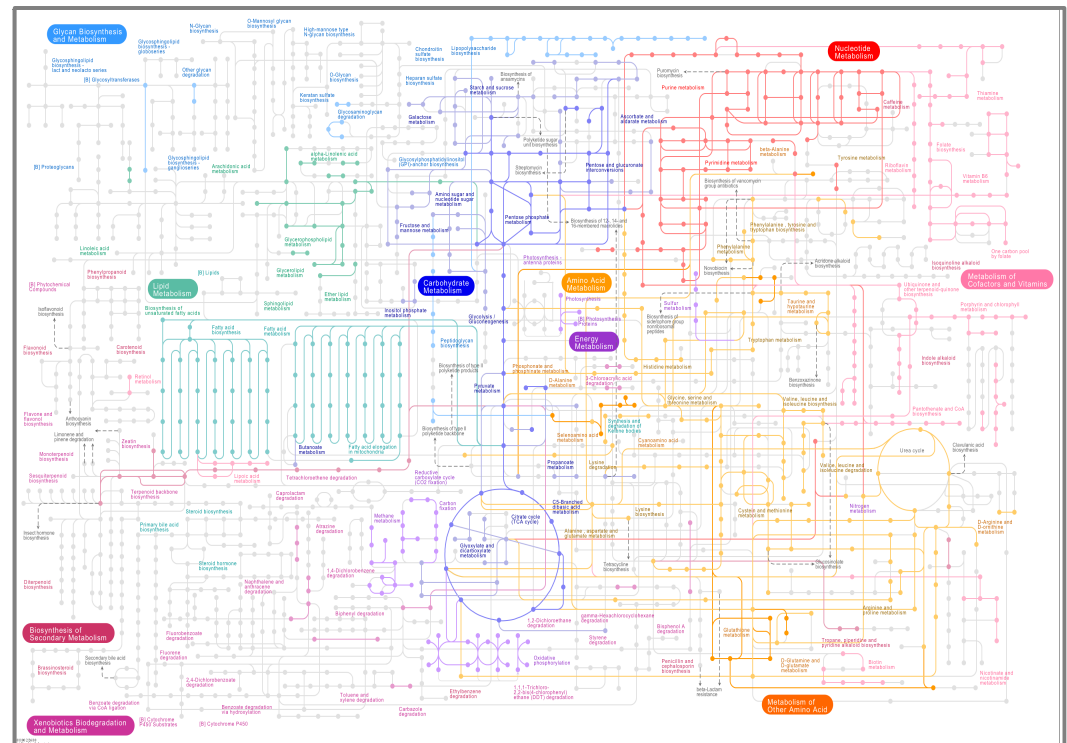
Genome-scale network models of *E. coli* metabolism

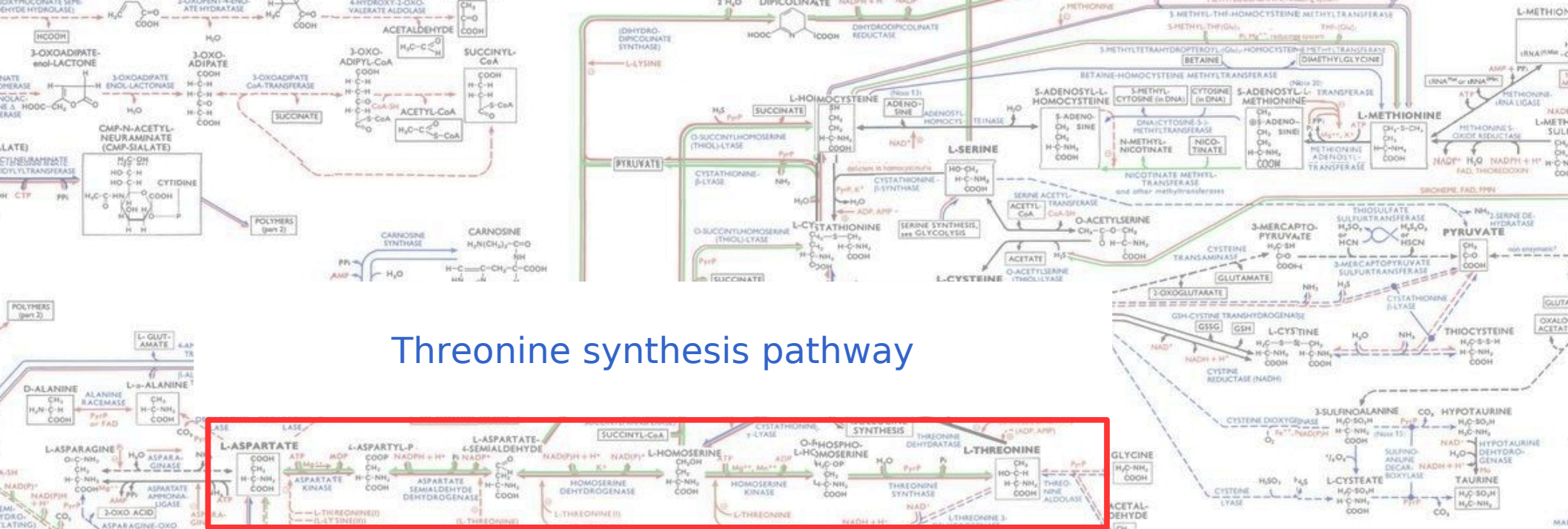
Molecular Systems Biology 3; Article number 121; doi:10.1038/msb4100155
Citation: *Molecular Systems Biology* 3:121
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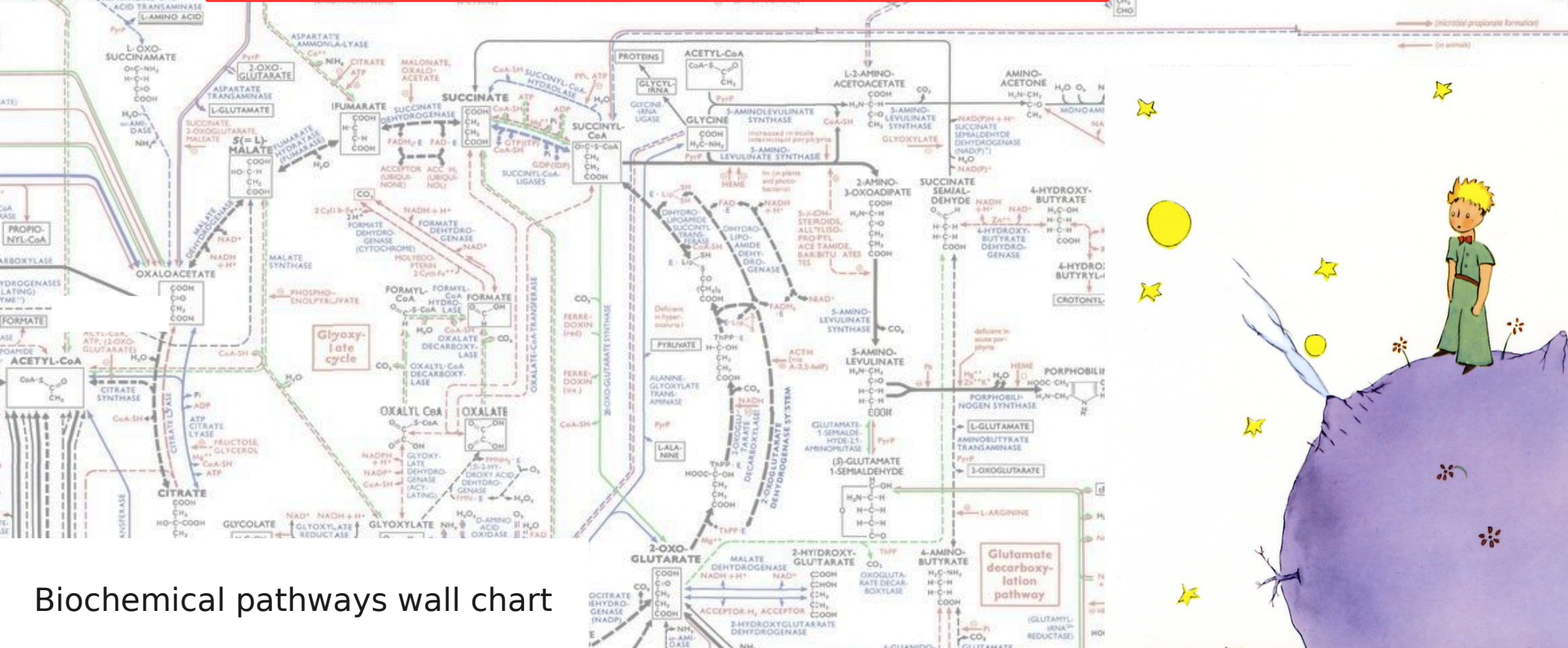
A genome-scale metabolic reconstruction for *Escherichia coli* K-12 MG1655 that accounts for 1260 ORFs and thermodynamic information

Adam M Feist¹, Christopher S Henry², Jennifer L Reed¹, Markus Krümhennacker³, Andrew R Joyce¹, Peter D Karp³,
Linda J Broadbelt², Vassily Hatzimanikatis¹ and Bernhard Ø Palsson^{1,*}





Threonine synthesis pathway



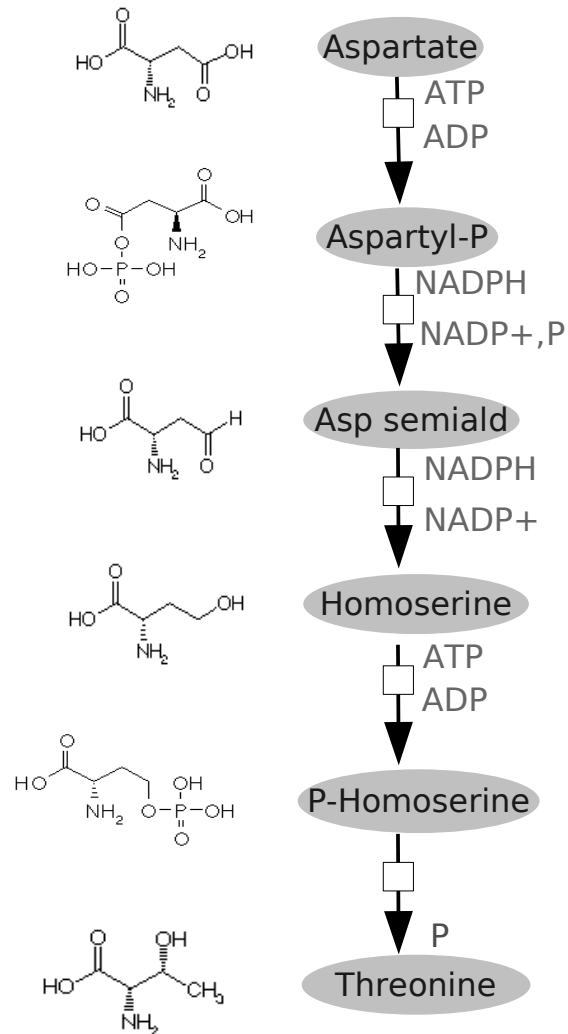
Biochemical pathways wall chart



Metabolic networks have several levels of regulation

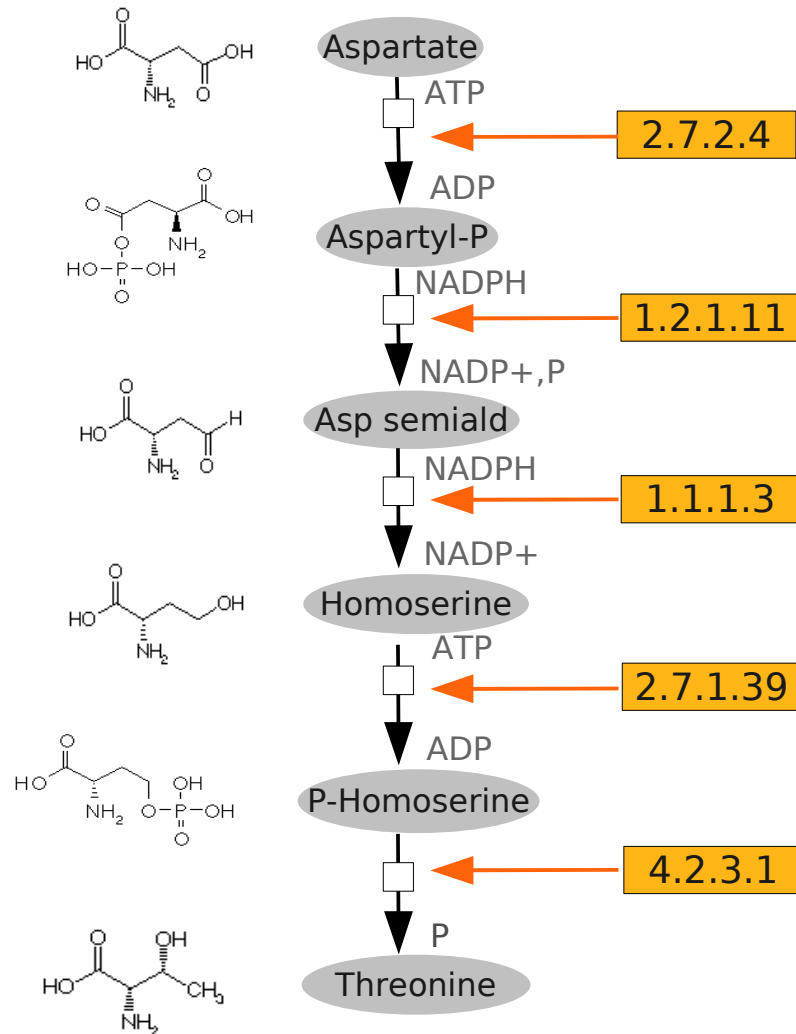
Metabolites

Reactions

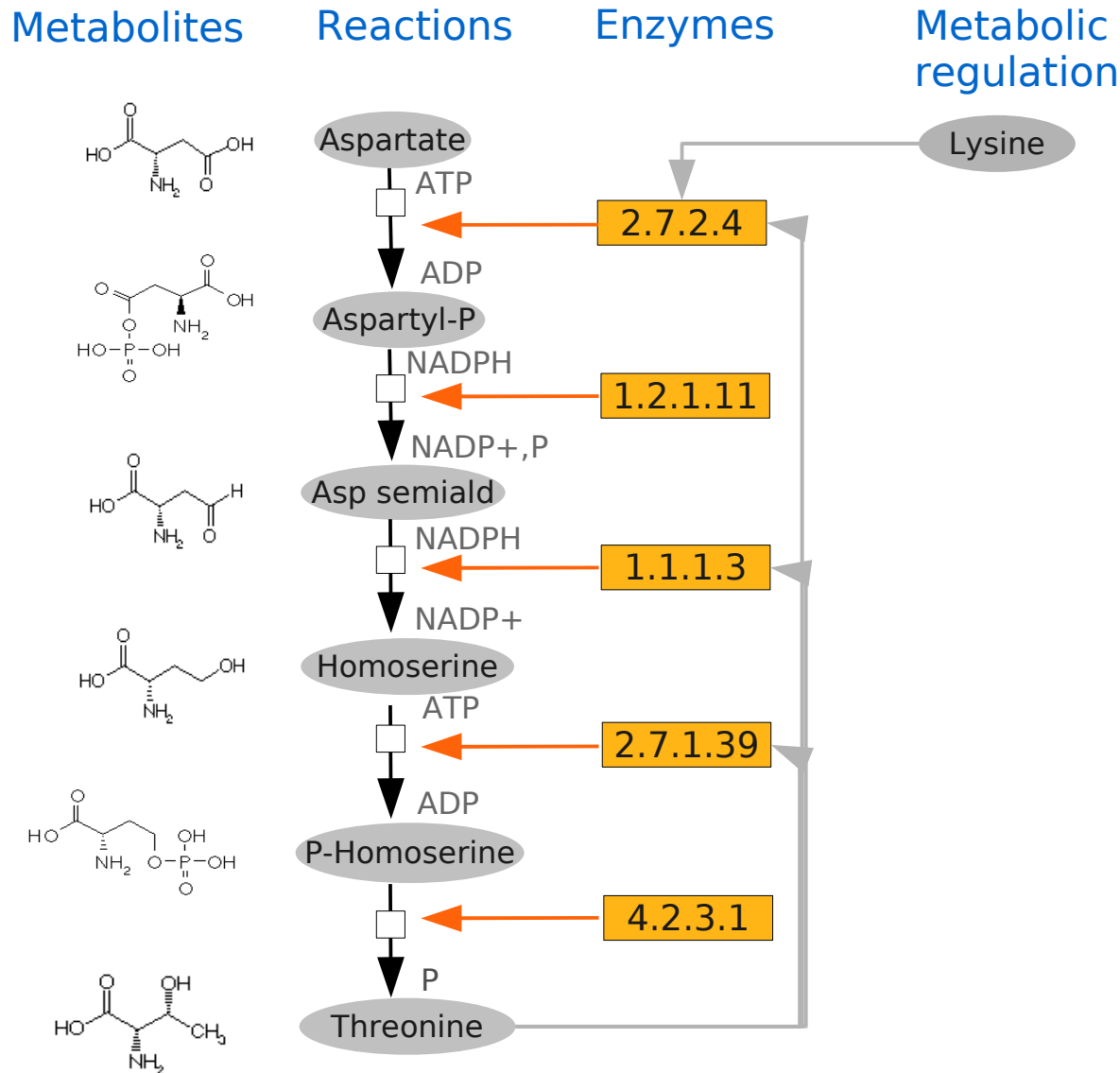


Metabolic networks have several levels of regulation

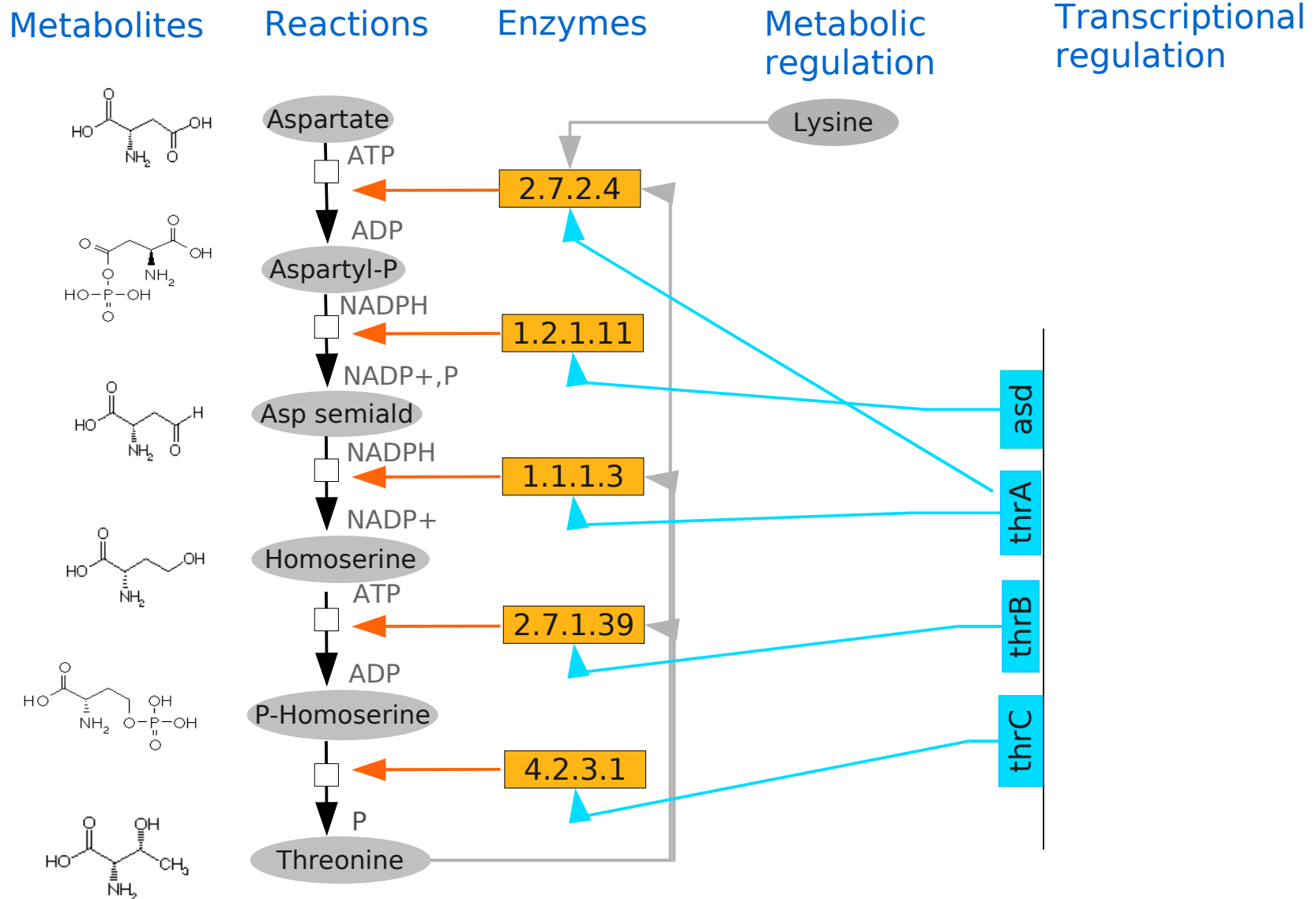
Metabolites Reactions Enzymes



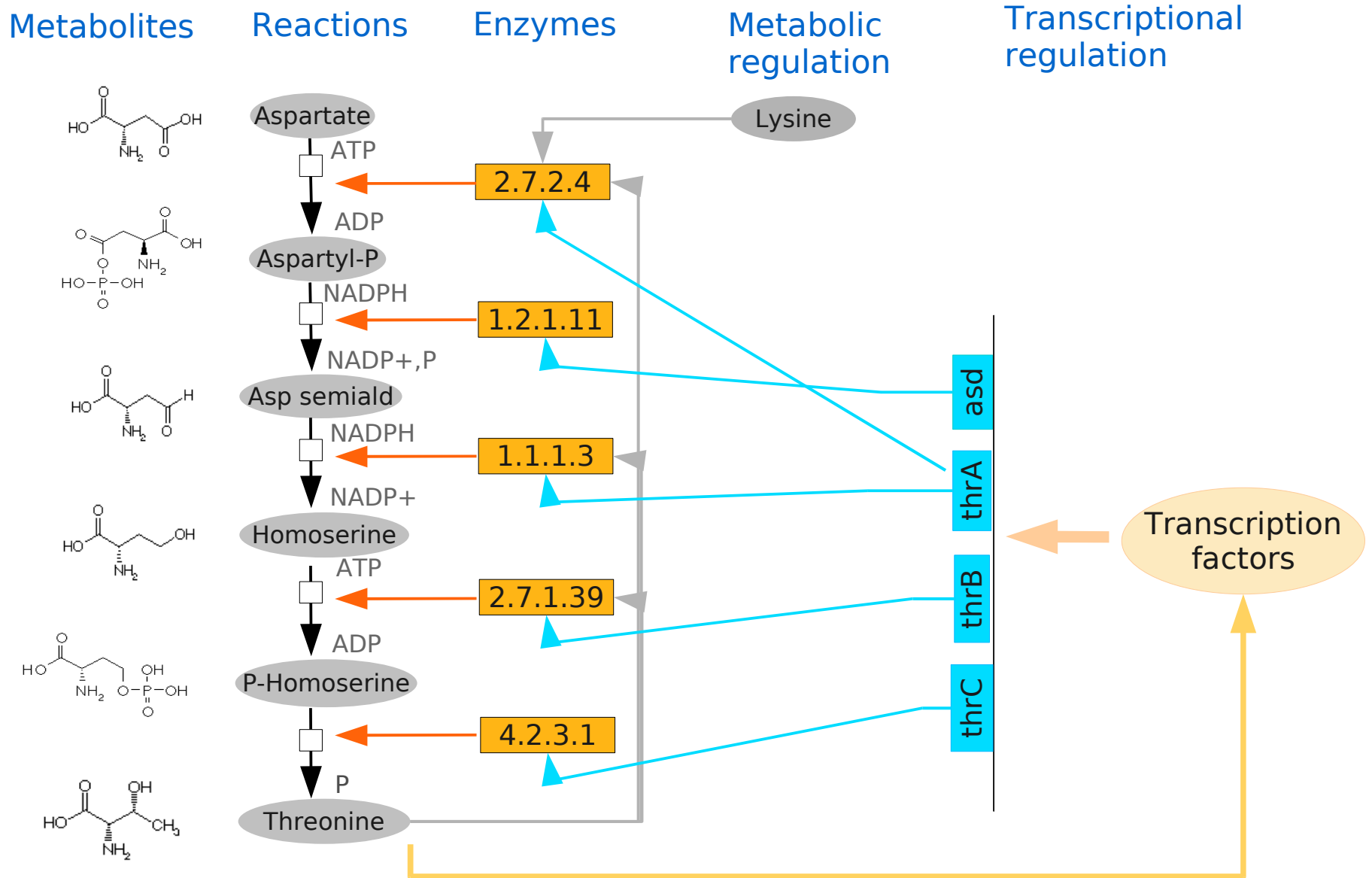
Metabolic networks have several levels of regulation



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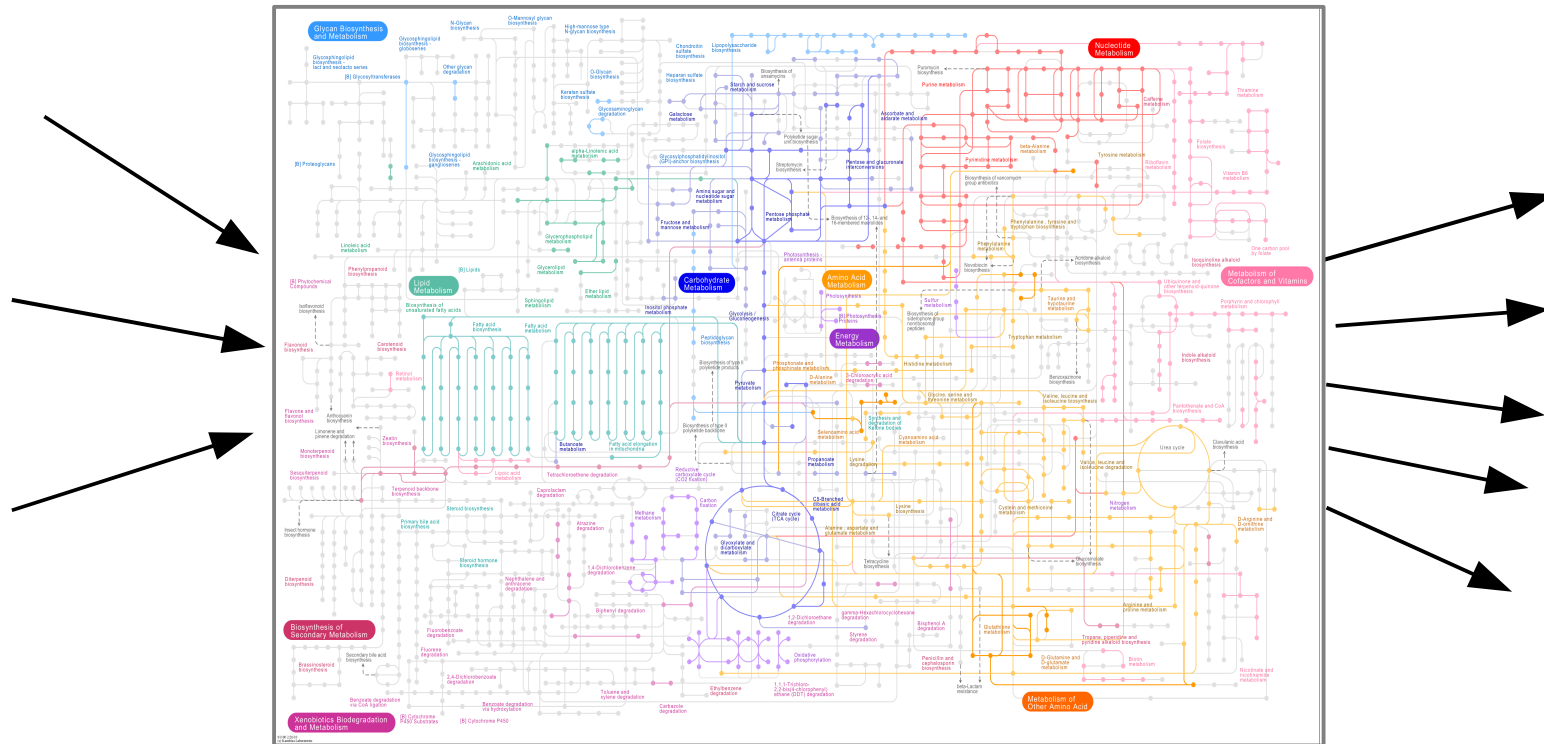


Metabolic networks have several levels of regulation



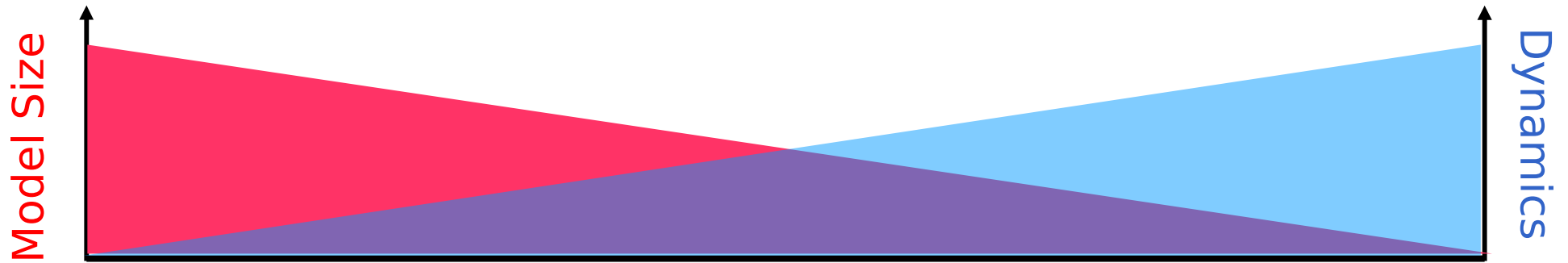
Kinetic models

How do metabolic networks work?

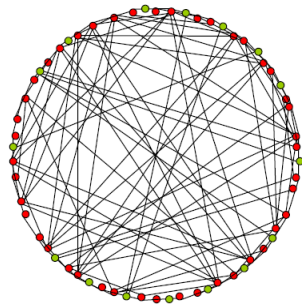


- What compounds can the cell produce?
- On which nutrient media can the cell survive?
- What do the metabolic fluxes look like ?
- How do they respond to varying conditions?
- How is all this regulated?
- What conclusions can we draw from limited data?

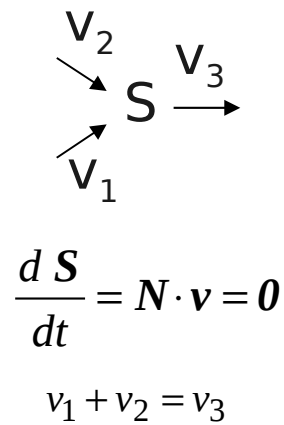
Modelling approaches for different complexity



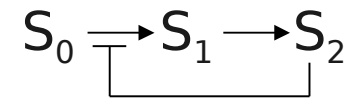
Topological Analysis



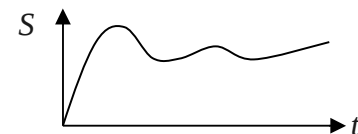
Flux Balance Analysis



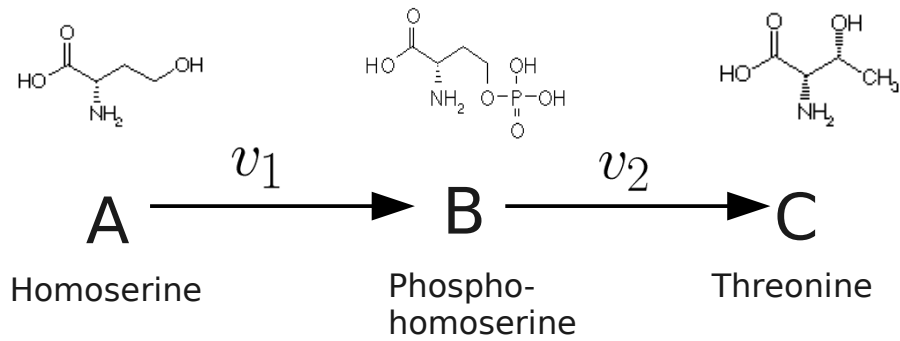
Kinetic modeling



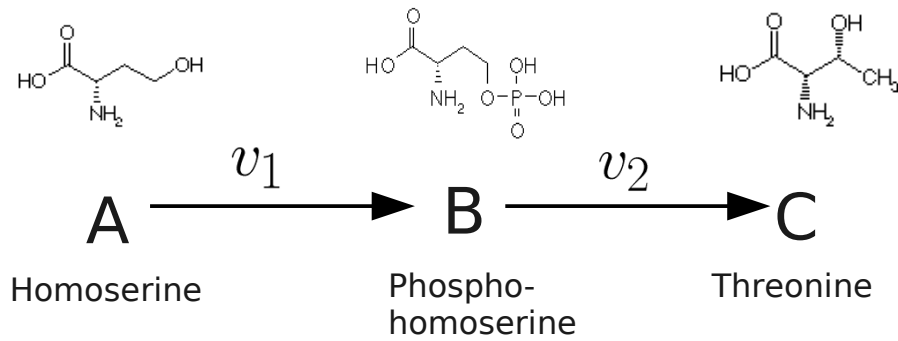
$$\frac{dS}{dt} = N \cdot v(S, p)$$



Kinetic models describe the dynamics of biochemical reactions



Kinetic models describe the dynamics of biochemical reactions

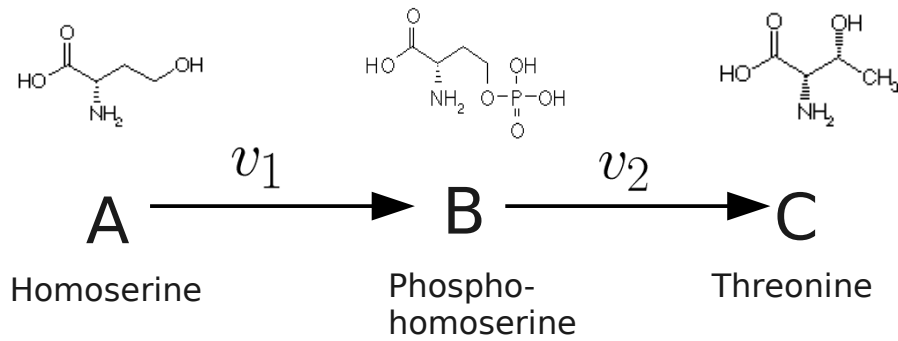


Reaction rate (“kinetic equations”)
How often does the reaction occur per time ?

$$v = k_+ a - k_- b$$

reaction rate kinetic constant concentration

Kinetic models describe the dynamics of biochemical reactions



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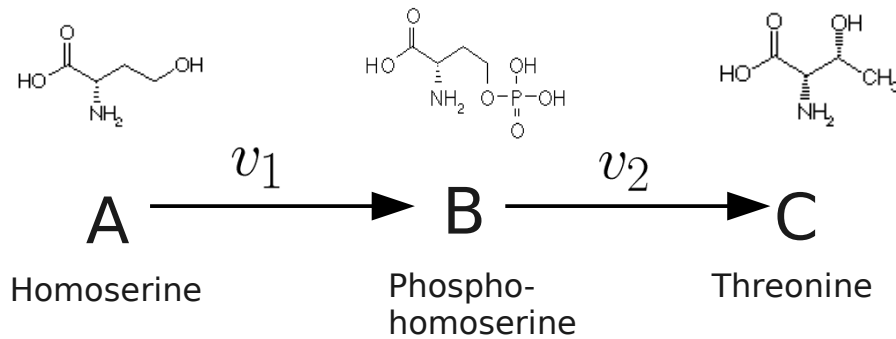
System equations
How do the concentrations change over time?

$$da/dt = -v_1$$

$$db/dt = v_1 - v_2$$

$$dc/dt = v_2$$

Kinetic models describe the dynamics of biochemical reactions



Reaction rate (“kinetic equations”)
How often does the reaction occur per time ?

$$v = k_+ a - k_- b$$

reaction rate \uparrow v
 kinetic constant \uparrow k_+
 concentration \uparrow a and b

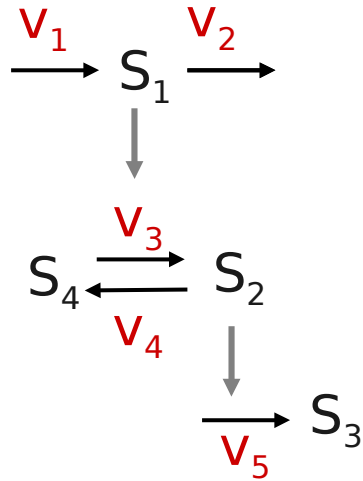
$$\frac{dx_i}{dt} = \sum_l n_{il} v_l(x, p)$$

concentration change \downarrow $\frac{dx_i}{dt}$
 stoichiometric coefficient \downarrow n_{il}
 kinetic law for reaction velocity \downarrow $v_l(x, p)$
 concentrations \uparrow x
 kinetic parameters and enzyme concentrations \uparrow p

System equations
How do the concentrations change over time?

$$\begin{aligned}
 da/dt &= -v_1 \\
 db/dt &= v_1 - v_2 \\
 dc/dt &= v_2
 \end{aligned}$$

System equations - an example



Metabolite
Concentrations

$$\vec{s} = \begin{pmatrix} S_1 \\ S_2 \\ S_3 \\ S_4 \end{pmatrix}$$

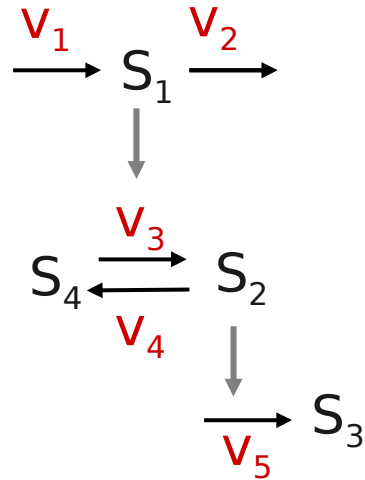
Reaction rates

$$\vec{v} = \begin{pmatrix} v_1 \\ v_2 \\ v_3 \\ v_4 \\ v_5 \end{pmatrix}$$

Stoichiometric Matrix

$$\mathbf{N} = \begin{matrix} & v_1 & v_2 & v_3 & v_4 & v_5 \\ \begin{pmatrix} 1 & -1 & 0 & 0 & 0 \\ 0 & 0 & 1 & -1 & 0 \\ 0 & 0 & 0 & 0 & 1 \\ 0 & 0 & -1 & 1 & 0 \end{pmatrix} & S_1 \\ & S_2 \\ & S_3 \\ & S_4 \end{matrix}$$

System equations - an example



Metabolite Concentrations

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ODEs

$$d[S_1]/dt = v_1 - v_2$$

$$d[S_2]/dt = v_3 - v_4$$

$$d[S_3]/dt = v_5$$

$$d[S_4]/dt = -v_3 + v_4$$

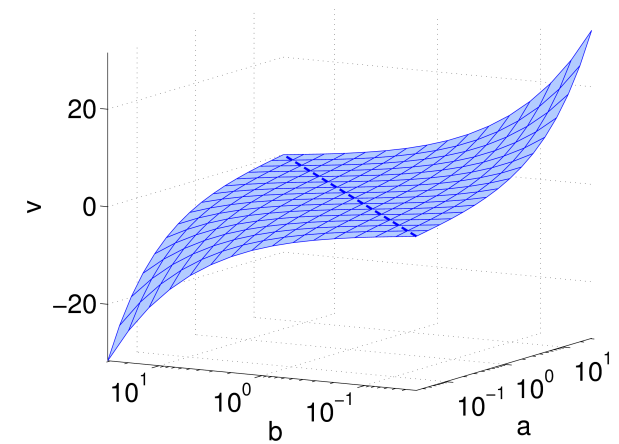
$$N \times \vec{v} = \vec{d[S]/dt}$$

$$\begin{pmatrix} 1 & -1 & 0 & 0 & 0 \\ 0 & 0 & 1 & -1 & 0 \\ 0 & 0 & 0 & 0 & 1 \\ 0 & 0 & -1 & 1 & 0 \end{pmatrix} \times \begin{pmatrix} v_1 \\ v_2 \\ v_3 \\ v_4 \\ v_5 \end{pmatrix} = \begin{pmatrix} v_1 - v_2 + 0 + 0 + 0 \\ 0 + 0 + v_3 - v_4 + 0 \\ 0 + 0 + 0 + 0 + v_5 \\ 0 + 0 - v_3 + v_4 + 0 \end{pmatrix}$$

The big problem in kinetic modelling: each enzyme is different !!

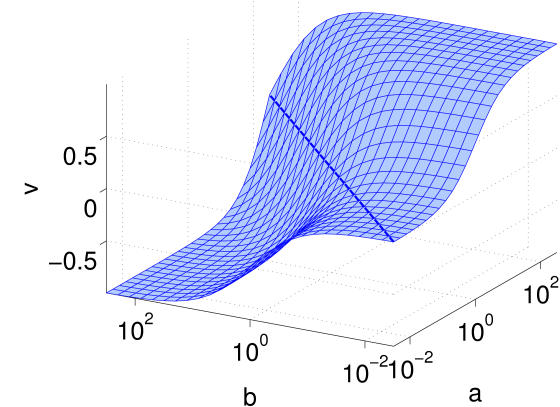
Mass-action kinetics (non-enzymatic reactions)

$$v = k_+ a - k_- b$$



Michaelis-Menten kinetics (simple enzymatic law)

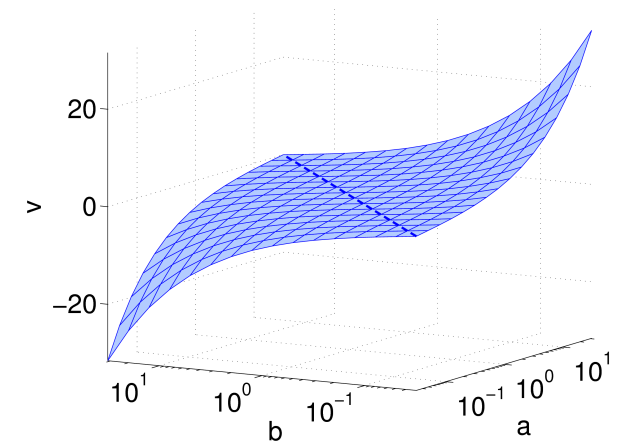
$$v = \frac{v_+^{\max}(a/k_A^M) - v_-^{\max}(b/k_B^M)}{1 + (a/k_A^M) + (b/k_B^M)}$$



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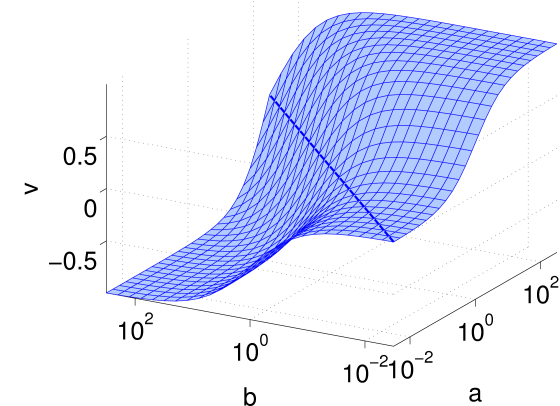
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Thermodynamics helps to reduce unknown parameters

Chemical equilibrium

$$0 = v(a^{\text{eq}}, b^{\text{eq}}) = v_+^{\max} \frac{a^{\text{eq}}}{k_A^M} - v_-^{\max} \frac{b^{\text{eq}}}{k_B^M}$$

Haldane relation

$$k^{\text{eq}} = \frac{b^{\text{eq}}}{a^{\text{eq}}} = \frac{v_+^{\max} k_B^M}{v_-^{\max} k_A^M}$$

Constraint-based models predict metabolic fluxes in large networks

External metabolites (e.g. extracellular or buffered)

Treated as fixed parameters

Intracellular metabolites (dynamic)

Concentration changes due to chemical reactions

Stationary (=steady) state

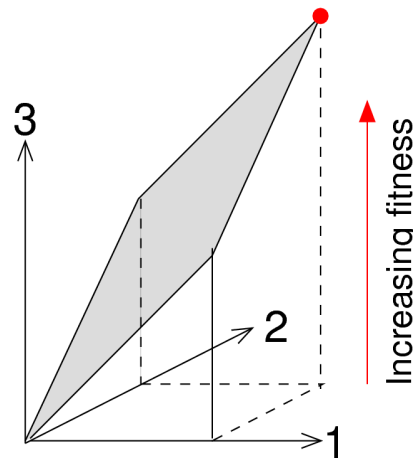
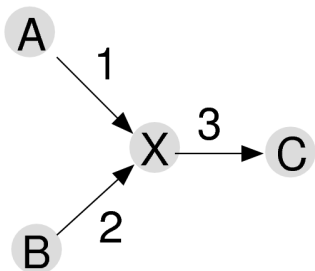
A state in which all variables remain constant in time

Stationarity condition in kinetic models

$$\frac{dc}{dt} = Nv = 0$$

Condition on the flux vector

Kinetic rate laws do not play a role!



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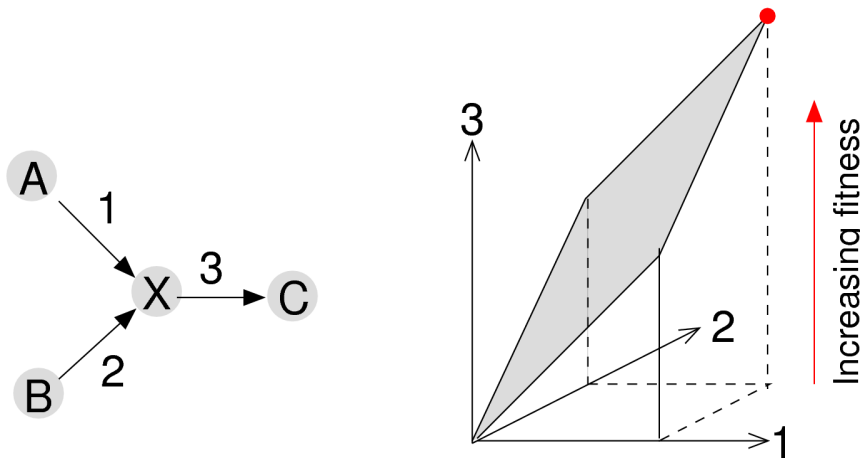
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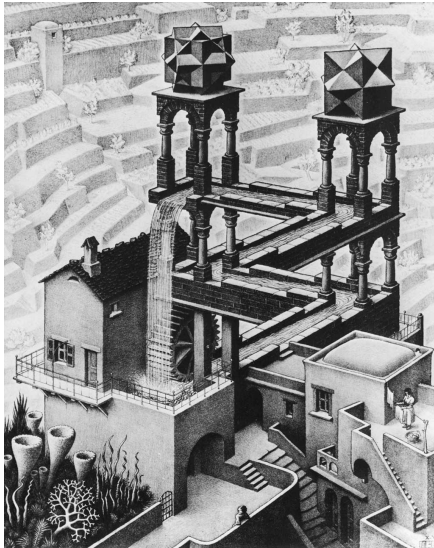
Flux balance analysis
predicts flux distributions for large networks

Stationarity + Upper and lower bounds on fluxes
→ Convex set in flux space

Linear optimisation (e.g. maximal product yield)
→ Linear programming problem

Fluxes have to satisfy thermodynamic constraints

1. Wegscheider conditions



$$f = -\nabla\Phi \quad \Rightarrow \quad \oint f(s) \cdot ds = 0$$
$$\Delta x = N^T x \quad \Rightarrow \quad K^T \Delta x = 0 \quad (\text{where } NK = 0)$$

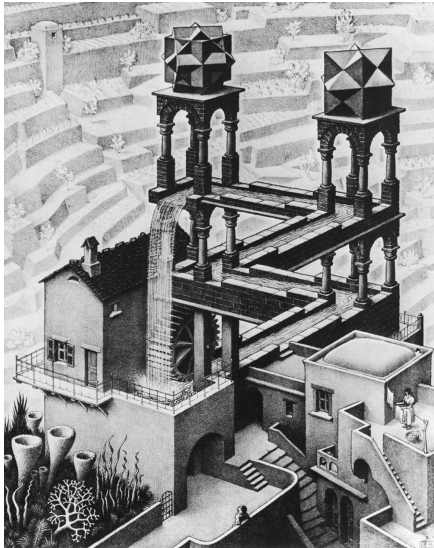
Equilibrium constants $K^T \ln k^{\text{eq}} = 0$

Mass-action ratios $K^T \ln q^{\text{ma}} = 0$

Reaction affinities $K^T A = -K^T \Delta\mu = 0$

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2. Flux directions and affinities (positive entropy production !)

$$v_r \neq 0 \quad \Rightarrow \quad \text{sign}(v_r) = \text{sign}(A_r) = -\text{sign}(\Delta\mu_r)$$

Metabolic control analysis traces the global effects of local changes

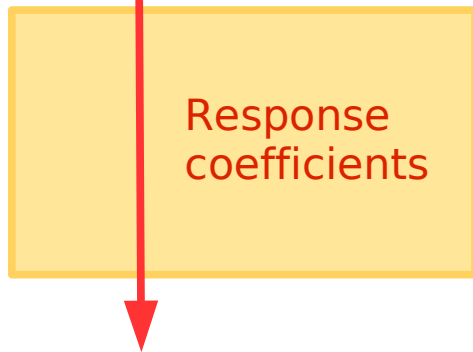
Parameter change
higher substrate supply? Δp_m



Metabolic change
altered concentrations?
redirected fluxes? $\Delta s_i \approx R_{p_m}^{s_i} \Delta p_m$

Metabolic control analysis traces the global effects of local changes

Parameter change
higher substrate supply? Δp_m



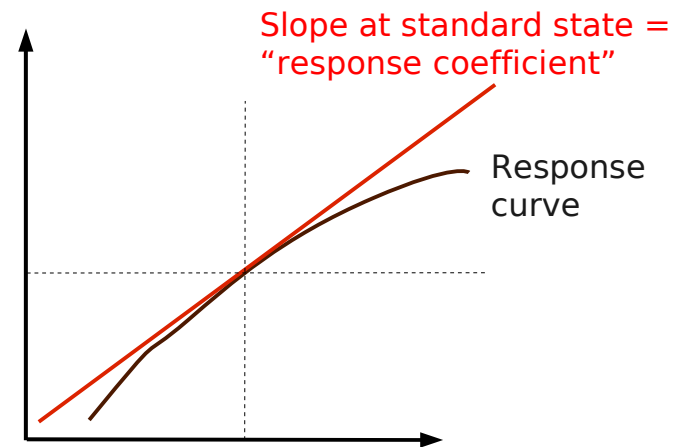
Metabolic change
altered concentrations?
redirected fluxes? $\Delta s_i \approx R_{p_m}^{s_i} \Delta p_m$

1. Stationary concentrations $s(p)$

Solution of $0 = N v(s(p), p)$

2. Response coefficients

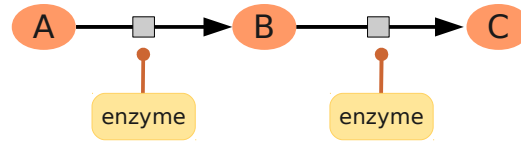
Systemic effect:
flux or concentration



Local cause:
e.g., single enzyme level

Summary: Modelling formalisms for biochemical systems

Kinetic models



$$\frac{dx_i}{dt} = \sum_l n_{il} v_l(x, p)$$

concentration

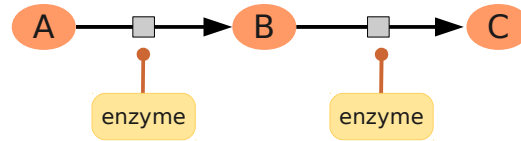
stoichiometry

reaction rate

parameters

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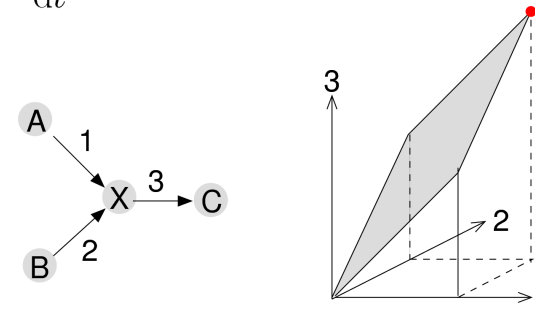


$$\frac{dx_i}{dt} = \sum_l n_{il} v_l(x, p)$$

concentration $\frac{dx_i}{dt}$
stoichiometry n_{il}
reaction rate $v_l(x, p)$
parameters p

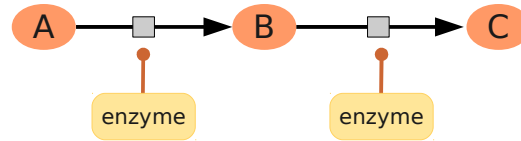
Constraint-based models (e.g., flux balance analysis)

$$\frac{dx}{dt} = 0 \Rightarrow Nv = 0$$



Summary: Modelling formalisms for biochemical systems

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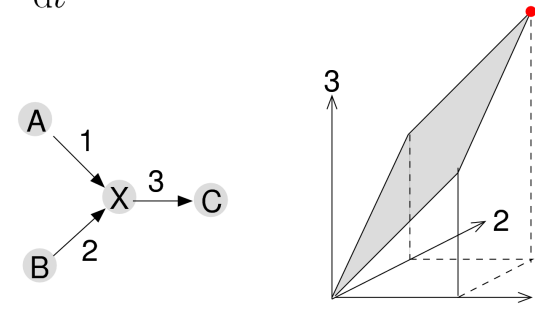


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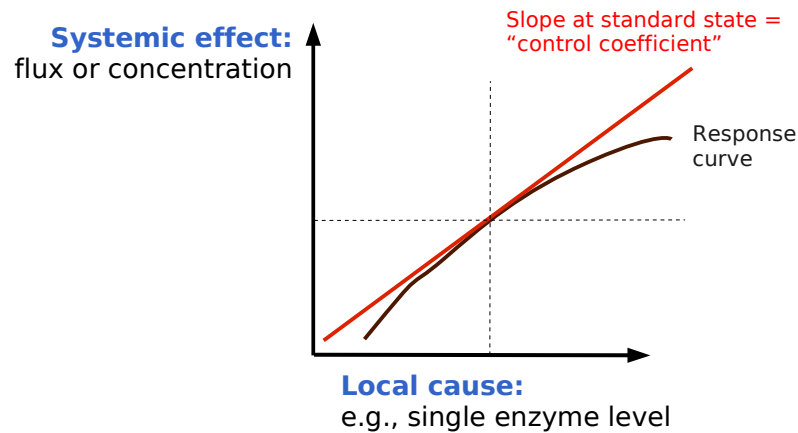
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 stoichiometry $\rightarrow n_{il}$
 reaction rate $\rightarrow v_l(x, p)$
 parameters $\rightarrow p$

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$$\frac{dx}{dt} = 0 \Rightarrow N v = 0$$

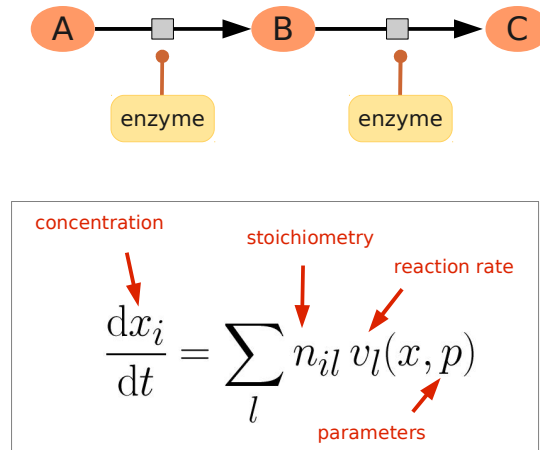


Metabolic control theory

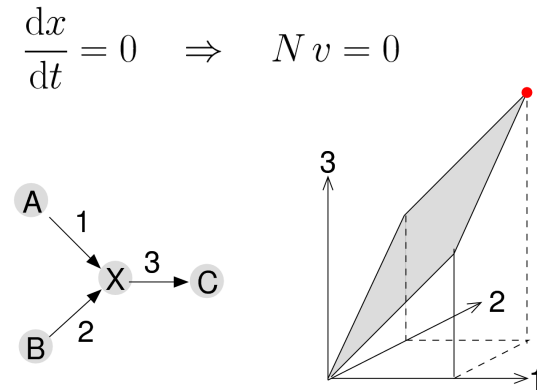


Summary: Modelling formalisms for biochemical systems

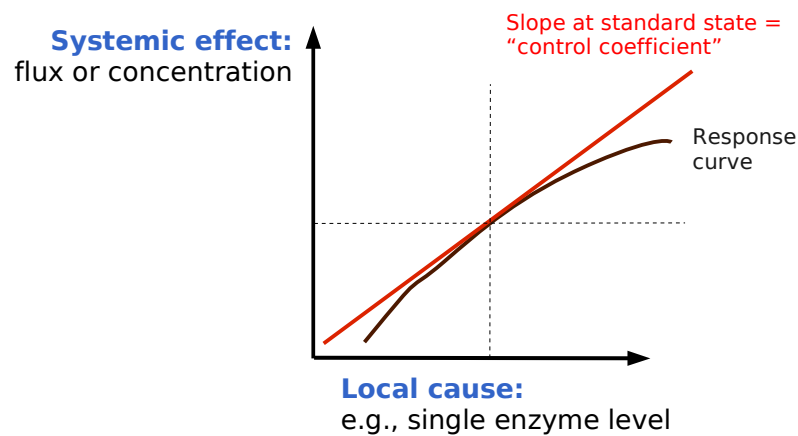
Kinetic models



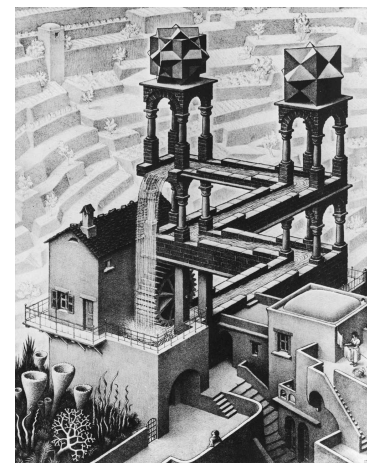
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Metabolic control theory



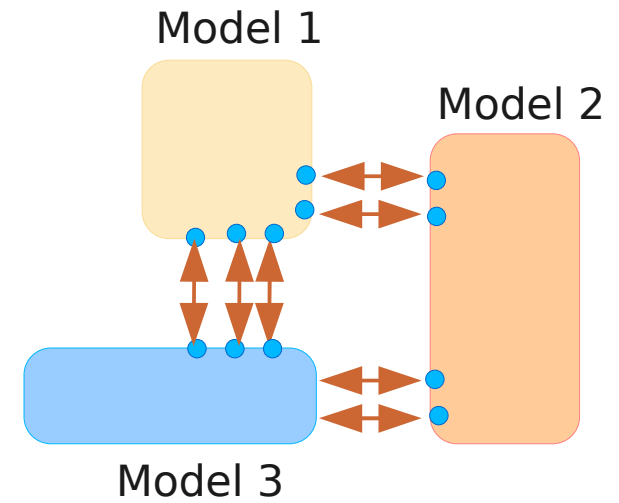
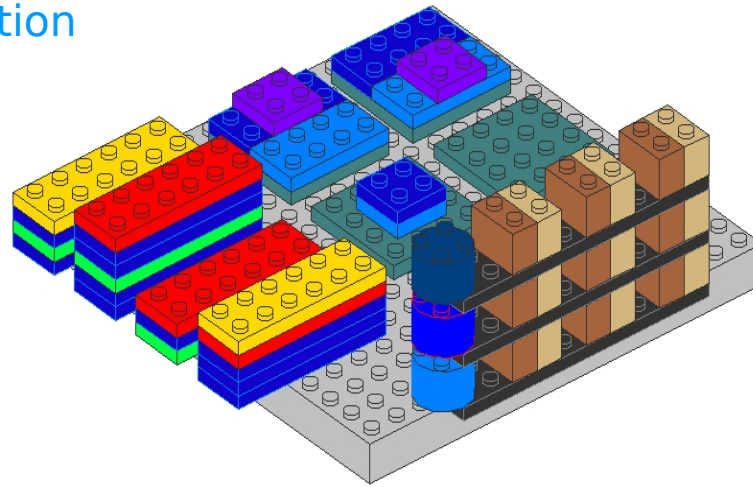
Thermodynamic analysis



Technical resources for modelling

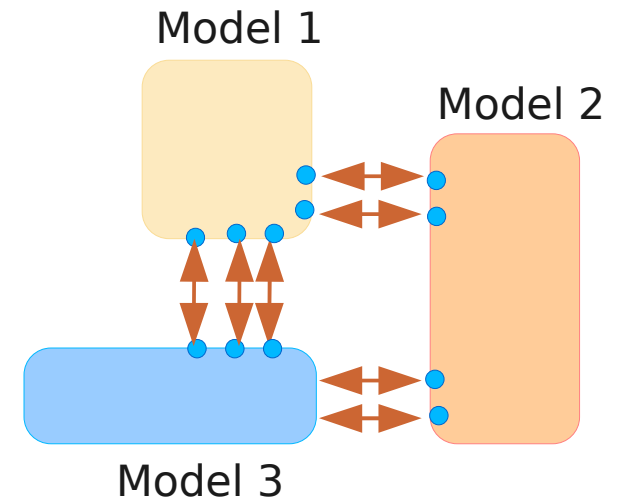
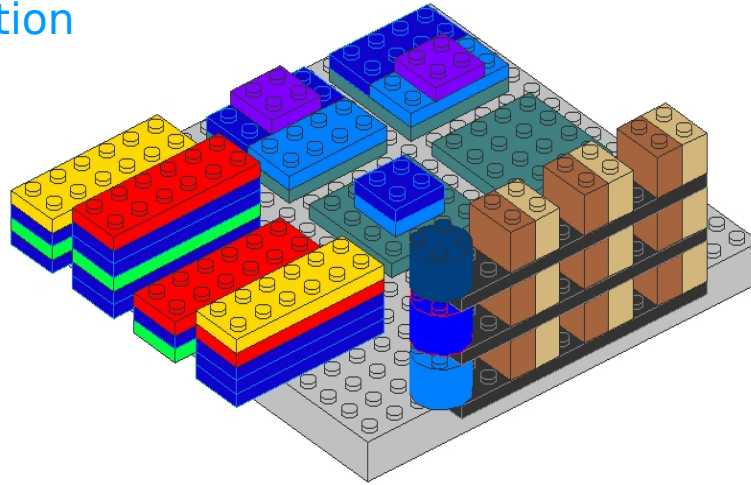
Playing with biochemical models ?

Model composition

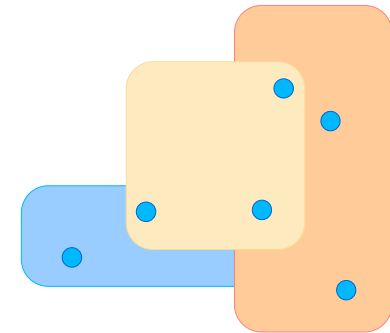


Playing with biochemical models ?

Model composition



Model merging

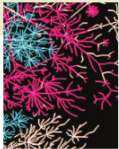


Models should be reusable



“Most of the published quantitative models in biology are lost for the community because they are either not made available or they are insufficiently characterized to allow them to be reused.”

Le Novère *et al*, (2005)



computational
BIOLOGY

PERSPECTIVE

Minimum information requested in the annotation of biochemical models (MIRIAM)

Nicolas Le Novère^{1,15}, Andrew Finney^{2,15}, Michael Hucka³, Upinder S Bhalla⁴, Fabien Campagne⁵, Julio Collado-Vides⁶, Edmund J Crampin⁷, Matt Halstead⁷, Edda Klipp⁸, Pedro Mendes⁹, Poul Nielsen⁷, Herbert Sauro¹⁰, Bruce Shapiro¹¹, Jacky L Snoep¹², Hugh D Spence¹³ & Barry L Wanner¹⁴

Most of the published quantitative models in biology are lost for the community because they are either not made available or they are insufficiently characterized to allow them to be reused. The lack of a standard description format, lack of stringent reviewing and authors' carelessness are the main causes for incomplete model descriptions. With today's increased interest in detailed biochemical models, it is necessary to define a minimum quality standard for the encoding of those models. We propose a set of rules for curating quantitative models of biological systems. These rules define procedures for encoding and annotating models represented in machine-readable form. We believe their application will enable users to (i) have confidence that curated models are an accurate reflection of their associated reference descriptions, (ii) search collections of curated models with precision, (iii) quickly identify the biological phenomena that a given curated model or model constituent represents and (iv) facilitate model reuse and composition into large subcellular models.

During the genomic era we have witnessed a vast increase in availability of large amounts of quantitative data. This is motivating a shift in the focus of molecular and cellular research from qualitative descriptions of biochemical interactions towards the quantification of such interactions and their dynamics. One of the tenets of systems biology is the use of quantitative models (see **Box 1** for definitions) as a mechanism for capturing precise hypotheses and making predictions^{1,2}. Many specialized models exist that attempt to explain aspects of the cellular machinery. However, as has happened with other types of biological information, such as sequences, macromolecular structures or

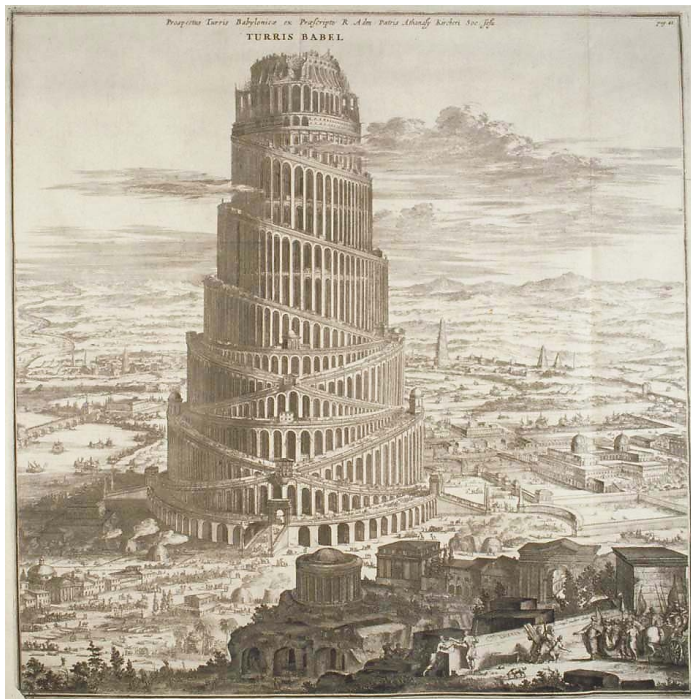
Box 1 Glossary

Some terms are used in a very specific way throughout the article. We provide here a precise definition of each one.

Quantitative biochemical model. A formal model of a biological system, based on the mathematical description of its molecular and cellular components, and the interactions between those components.

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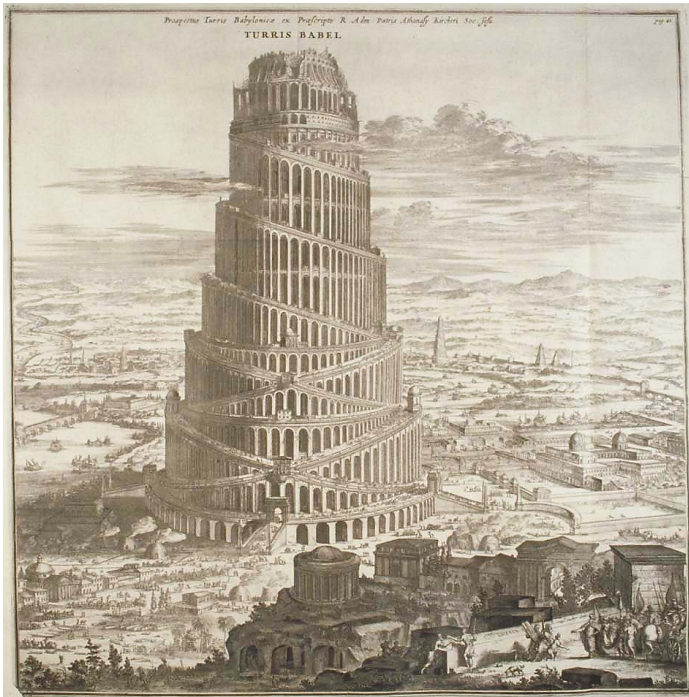
Systems Biology Markup Language (SBML)



Systems Biology Markup Language (SBML)



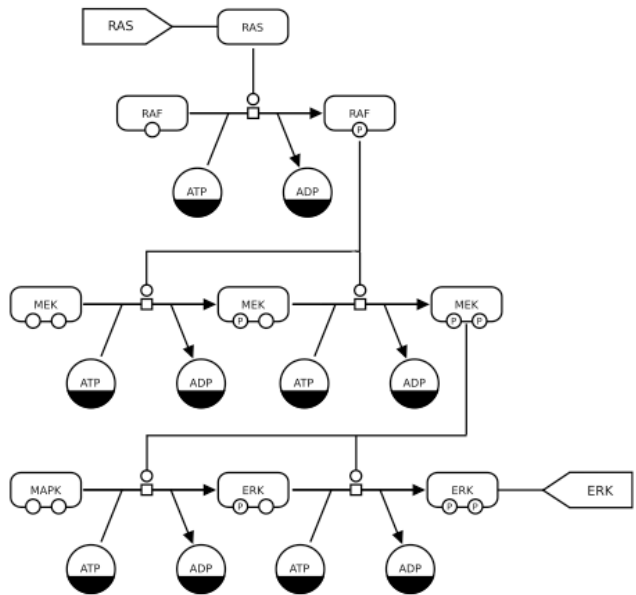
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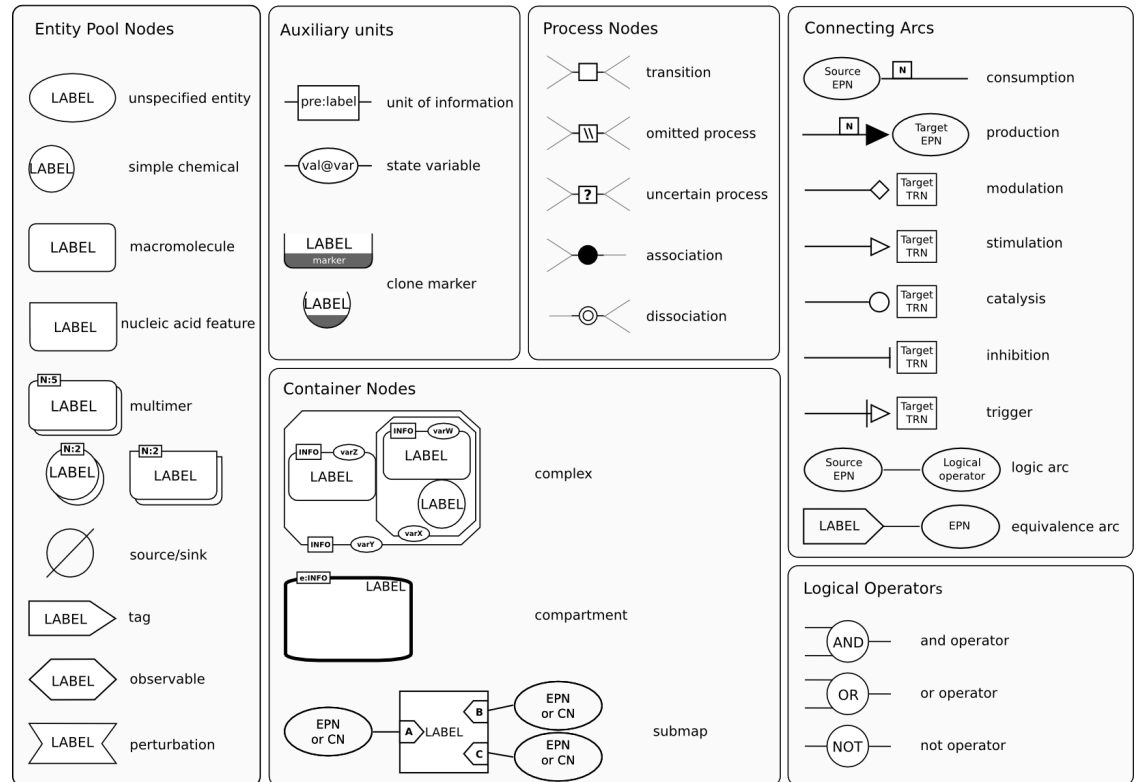
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Systems Biology Graphical Notation (SBGN)

Process description diagram



SYSTEMS BIOLOGY GRAPHICAL NOTATION REFERENCE CARD



Data, modelling software, and models are available on the web

Network reconstructions



Databases for biological numbers

BRENDA

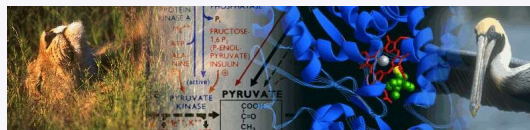


B10NUMB3R5
THE DATABASE OF USEFUL BIOLOGICAL NUMBERS

Model repositories



Database of curated annotated models
<http://biomodels.org/>



JWS online: database of curated models
<http://jjj.biochem.sun.ac.za/>

Modelling software

<http://sbml.org/>



SB.OS – Live DVD with free modelling software

SB.OS / Systems Biology Operational Software

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Welcome to SB.OS - Systems Biology Operational Software

SB.OS is a live DVD based on Ubuntu Linux that comes with a comprehensive list of Systems Biology Software. Text and video documentation material, as well as an offline copy of the BioModels.net database, are included.

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You can run SB.OS on your computer without having to install anything. To create a bootable DVD just burn the disk image with your favourite burning software to a DVD. To create a bootable USB stick use UNetbootin together with our disk image. All you have to do now is to change the boot device of your computer to first boot from your DVD drive (this is the default on most computers) or to USB. Insert the DVD/USB stick and boot your computer.

start1

Software on SB.OS

- Included (version 3.0)
- Pending
- Excluded

About SB.OS

SB.OS Documentation

Team

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- Theoretical Biophysics

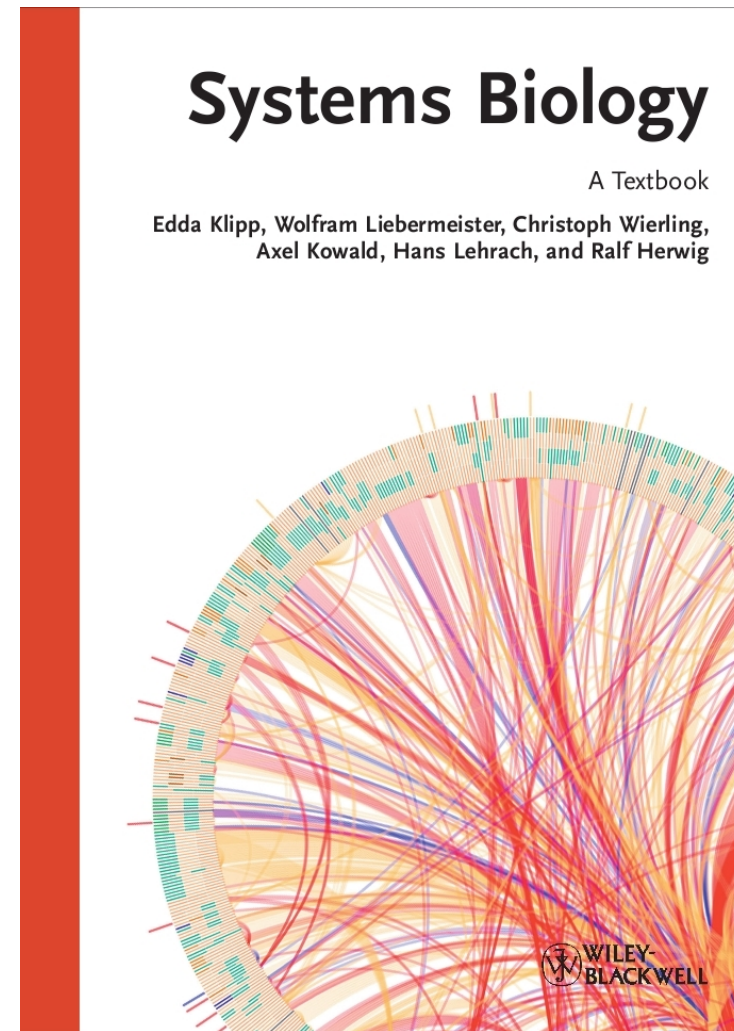
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